# **Critical Minerals, Recoverable from Radioactive Materials for Value-Addition: Indian Examples**

## R. Dhana Raju

Former Associate Director, AMD, Dept. of Atomic Energy, Govt. of India, Hyderabad – 500 016 Former Hon. Visiting Prof., Dept. of Applied Geochemistry, Osmania University, Hyderabad – 500 007 1-10-284/1, Brahmanwadi Lane 5, Begumpet, Hyderabad-500 016 *Email: dhanaraju.reddi@gmail.com* 

*Abstract*— Critical Minerals (CMs) are mostly metals and a few non-metals, which have diverse, important industrial uses and, thus, are vital for the economic well-being of the Society, but their supply may be at risk due to several factors. They occur in three sources, viz., primary, from ore deposits, recoverable mostly as by-products and less as main products; secondary, extractable from waste materials like the electronic waste and tertiary, in imports. The list of CMs is dynamic and depends upon their availability and perspective of each country, and, hence, may change with time. For India, 33 CMs were proposed earlier by the author— Sb, As, Be, Bi, Cs, Cr, Co, Ga, Au, C (natural), Hf, He, In, Li, Ni, Nb-Ta, PGMs, REEs, Re, Rb, Sc, Se, Ag, Sr, Te, Sn, Ti, W, U, V and Zr, and potash, and phosphate (proposed now, making the total 34), required for the fertilizer industry. Of these, > 60% CMs may be recoverable from radioactive materials (RMs). Examples of these with their contents in diverse RMs from various parts of India are presented and discussed. From such RMs, the CMs-carrying minerals are to be first recovered and concentrated by physical beneficiation techniques of chemical-/biochemical-/nano-technology, for which extensive R&D is to be carried out. By this swadeshi (indigenous) way under the Prime Minister's Make in India programme, the Country's requirement of critical/strategic minerals can be met to a notable extent, besides value-addition during the processing of RMs for their main products.

Keywords— Critical Minerals, Radioactive Materials, Value-Addition, Indian Examples.

### **INTRODUCTION**

Critical Minerals (CMs) are mostly metals and a few non-metals, which are vital for the economic well-being of the world's major and emerging e-economies. But their supply may be at risk due to geological scarcity, geopolitical issues, trade policy, supply-chain problems and other factors. CMs ranked as most critical for the world's major industrial economies due to their use in futuristic developments in energy, health, construction and transportation sectors as well as in space, nuclear and defence industries, and artificial intelligence of the developed countries, such as the USA, Japan, Republic of Korea, European Union (EU) and the UK include rare earth elements (REEs), platinum group metals (PGMs), lithium (Li), beryllium (Be), gallium (Ga), germanium (Ge), indium (In), tungsten (W), cobalt (Co), niobiumtantalum (Nb-Ta), molybdenum (Mo), antimony (Sb), vanadium (V), nickel (Ni), tellurium (Te), chromium (Cr), tin (Sn), thorium-uranium (Th-U), zirconium (Zr), hafnium (Hf), selenium (Se), rhenium (Re), phosphate, potash, etc. (Critical Minerals, Geoscience Australia). From the perspective of each country, the list of CMs may change. For example, USA lists 35 CMs in the year, 2018 (Dept. of the Interior, US Geological Survey, 2018) and increased to 54 CMs in their draft for the year 2021, based on the lowest supply chain risk and

quantitative evaluation (Federal Register, 2021: Ga, Nb, Co, Nd, Ru, Rh, Dy, Al, fluorspar, Pt, Ir, Pm, Ce, La, Bi, Y, Sb, Ta, Hf, W, V, Sn, Mg, Ge, Pd, Ti, Zn, graphite, Cr, As, barite, In, Sm, Mn, Li, Te, Pb, potash, Sr, Rh, Ni, Cu, Be, feldspar, phosphate, Ag, mica, Se, Cd, Zr, Mo, Au, He and, iron ore [54]; besides 12, based on qualitative evaluation - Cs, Er, Gu, Gd, Ho, Lu, Rb, Sc, Te, Th, U and Yb), as these are considered critical to its economic and national security (President of the United States, Executive order13817, 2018; US Geological Survey, 2018). For Australia, its critical commodities include Sb, Be, C (graphite), He, In, Li, Mn, Mo, Nb-Ta, Th, Sn, Ti, W and Zr, besides phosphate and potash used in fertilizers (Critical Minerals, Geoscience Australia) and this list is increased to 27 comprising two categories: High Criticality: REE, Ga, In, W, PGE, Co, Ni, Mg, Mo, Sb, Li, V, Ni, Ta Te, Cr and Mn; Moderate Criticality: Se, Ti, Sr, graphite, Sn, Ge, Be, Zr, Bi and F) (Australian Government, Dept. of Industry, Science, Energy and Resources). In India, 10 strategic minerals (S, Pb, petroleum, Zn, Hg, Pt, Ni, graphite, Sn and ferrotungsten) have been identified and their present status and future challenges are discussed by Ranadive and Jawadan (2019). From the Indian perspective, the present author has earlier suggested 33 CMs (Dhana Raju, 2020) and now adds one (phosphate) more taking the total to 34, viz., antimony, arsenic, beryllium,

bismuth, caesium, chromium, cobalt, gallium, germanium, gold, graphite (natural), hafnium, helium, indium, lithium, nickel, niobium, phosphate, platinum group metals (PGMs), potash, rare earth elements (REEs), rhenium, rubidium, scandium, selenium, silver, strontium, tantalum, tellurium, tin, titanium, tungsten, uranium, vanadium and zirconium. Of these, over 60% CMs, due to their notable content, are recoverable from diverse radioactive materials (RMs) - rock, ore, mineral, sand/concentrate. mineral Examples of these recoverable CMs from the Indian RMs, their industrial uses and their contents are presented in this article, together and purification of CMs by appropriate R&D, under the Prime Minister's Make in India programme, so as to meet the Country's requirement of critical/strategic minerals to a notable extent, besides value-addition during the processing of RMs.

#### CRITICAL MINERALS<mark>, RECOVERABLE FR</mark>OM RADIOACTIVE MATERIALS, AND THEIR INDUSTRIAL USES

Radioactive Materials (RM) are diverse in nature and include rock, ore, mineral, mineral sand and mineral concentrate, with their radioactivity ascribed to the presence of radio-elements, viz., Uranium (U), Thorium(Th) and radio-isotope of potassium (40K), in them in concentrations much higher than their crustal average. Some of the Critical Minerals (CMs, cited above), due to their geochemical affinity, are associated with the radio-elements in RMs e.g., REEs with U and Th in radioactive minerals like uraninite, zircon, thorianite, monazite, allanite etc. If such radioactive minerals are in notable concentration in diverse RMs, some of the CMs in them may be recovered, first by separating/concentrating the constituent radioactive minerals by subjecting the RMs to physical beneficiation, using the gravity, magnetic and electromagnetic etc., techniques, and next the CMs in the separated/concentrated radioactive minerals can be extracted, enriched and purified by a combination of techniques of diverse methods such as hydro-/pyrometallurgy, chemical-/bio-technology, nano-/lasertechnology, smelting etc. The CMs that are recoverable from diverse RMs are broadly categorized under 7 heads, viz., REEs, Battery Metals, Precious Metals, Nuclear Metals, Alloy Metals, Toxic Metals and Fertilizer Commodities, and their important industrial uses are given in Table 1.

Tuble 1. Cruical Minerals (recoverable from Radioactive Materials) and their maustrial Uses		
Critical Minerals	Industrial Uses*	
Symbol & At. No.		
1. REEs:	La: in hybrid engines, H-storage batteries, green phosphors, metal alloys, laser crystals (Y-	
Lanathanum, La-57	La-fluoride); La-rich compounds in FCC catalysts, mainly for making low-octane fuel from	
Cerium, Ce-58,	heavy crude oil; La-stabilized zirconia in electronics.	
Prasodymium, Pr59	Ce: in glass & glass-polishing, phosphors (as a sensitizer), ceramics (in dental compositions	
Neodymium,Nd-60	and as a phase stabilizer in zirconia-based products), catalysts (FCC- catalysts, as a stabilizer	
Promythium,Pm-61	in catalytic converters, as a promoter of the water-gas shift reaction & as an O-storage	
Samarium, Sm-62	component), metallurgy, steel manufacturing (to remove free O/S by forming stable	
Europium, Eu-63	oxysulphides & by tying up undesirable trace elements); Ce-doped glass (to block out UV	
Gadolinium, Gd-64	light) in manufacturing medical glassware & aerospace windows, to prevent polymers from	
Terbium, Tb-65	darkening in sunlight & to suppress discoloration of television glass, and as optical	
Dysprosium, Dy-66	components to improve performance.	
Holium, Ho-67	Pr: in magnets, ceramics, pigment (bright yellow), in scintillator for medical computed	
Erbium, Er-68	tomography scans; Pr-doped zirconia.	
Thulium, Tm-69	Nd: in lasers (Nd-based YAG lasers in medical fields, drilling, welding & material	
Ytterbium, Yb-70	processing), glass colouring & tinting, dielectrics, electronic equipment, Nd2Fe14B	
Lutetium, Lu-71	permanent magnets (in HEVs, EVs & renewable energy generation), Y-Al-garnet solid-state	
Yttrium, Y-39 and	lasers, automotive industry, in the voice coil motors required for computer disk drives &, in	
Scandium, Sc-21	protective lenses for welding goggles.	
	Sm: in manufacturing of Sm-Co permanent magnets (replacing expensive Pt-Co magnets),	
	laser applications due to its dielectric properties (as stable Sm-titanate compounds for	
	coatings), in the filter glass on Nd: YAG solid state lasers (to envelope laser rod to improve	
	efficiency by absorbing stray emissions) & in capacitors at microwave frequencies.	
	Eu: has a unique luminescent property that is used for emission of visible radiation & in	
	medical, surgical & biochemical fields, energy-efficient fluorescent lighting (yields both red	
	& blue light), commercial red phosphors for color TV, computer screens & fluorescent	
	lamps.	

Table 1. Critical Minerals (recoverable from Radioactive Materials) and their Industrial Uses

	Gd: for both high magnetic moment and phosphors & scintillated materials, when mixed with EDTA dopants, it is used as an injectable contrast agent for patients undergoing magnetic resonance imaging, can reduce relaxation times (due to its high magnetic moment) and enhances signal intensity and as an inert phosphor host for x-ray cassettes & scintillated
	materials for computed tomography.
	Tb: used in phosphors, particularly in fluorescent lamps, & as high intensity green emitter,
	used in projection televisions (Tb: YAG), as an x-ray phosphor & Tb-alloys in magneto-
	optic recording films & magnets.
	Dy: mainly in Nd-Fe-B high strength permanent magnets & hybrid engines and in special ceramic compositions.
	Ho: Due to highest magnetic moment of any naturally occurring element, it is used to create
	the highest known magnetic fields by placing it within high strength magnets as a pole piece or magnetic flux concentrator, in YIronGarnet YIG) lasers for microwave equipment, in a variety of medical and dental fields in both YAG & YLF solid state lasers, applications in
	silica fibers designed for shorter wave-lengths, while still providing the cutting strength of longer wavelength equipment
	Fr: in phosphors, glass colouring, lasers for medical/dental use & in eve-ware decorative
	glassware, as amplifier for fiber optic data transfer, lasers (Er: YAG) for surgical applications
	The Most The products used in making errotals & locars for production of portable X row
	sources as tools for medical & dental diagnosis, for detecting defects in inaccessible
	mechanical & electronic components and in magnetic & ceramic materials (like Y-Fe alloys,
	required in the microwave technologies).
	Yb: in fiber amplifier/optic technologies & lasers, is useful in silicon photocells for
	converting directly radiant energy to electricity, as metal increases its electrical resistance on
	being subjected to very high stresses & used in stress gauges for monitoring ground
	deformations from earthquakes & nuclear explosions and as thermal barrier system bond coatings on Ni, Fe & other transition metal alloy substrates.
	Lu: is the best host for X-ray phosphors as it produces the densest known white material (Lu
<u></u>	tantalite), as a dopant in comparing crystallographic parameters of certain substrate garnet crystals (indium-gallium-garnet IGG) as it lacks in a magnetic moment
	$\mathbf{V}$ : (highest thermodynamic affinity for $\Omega$ ) in ceramics as crucibles for molten reactive
$\mathbf{\Lambda}$	metals, fluorescent lighting phosphors, computer displays & automotive fuel consumption sensors; Yttria-stabilized Zr-oxide in high temperature applica-tions (thermal plasma sprays
	for safety of aerospace component surfaces at high temperatures), YIG) crystals for the
	equipment of microwave communication &YAG crystals with Nd in a variety of laser
	applications; Yttria enhances the strength of metallic alloys; and Eu:Y2O2S phosphor
	produces red color in televisions.
	Sc: in ceramics, lasers, phosphors & certain high-performance alloys.
2. Battery Metals	Li: Primarily for Li-ion/-polymer batteries for electric-vehicles; in ceramics and glass
Lithium, Li-3	industries; nuclear fusion and for many chemical compounds.
Cobalt-27 and	Co: Co-oxide nano-particles in Li-ion batteries for electric vehicles and Co in super-alloys.
Nickel-23	Ni: For batteries, alloys, coins, cars, mobile phones, jet engines, cutlery and bathroom tops
	and shower heads
3 Precious Metals	PGMs: Pt catalysts in the production of silicones high-octane gas & petrochemical
PGMs. Platinum Pt_	feedstocks that are used to make plastics synthetic rubber & polyester fibres. Pt as ovygen
78 Palladium Pd_//6	sensors & snark plugs in automobiles. Pd in almost all electronics & in the manufacture of
$Osmium \qquad Os 76$	raw materials to make nylon & synthetic rubber because of its H absorbing qualities. Dt $k$
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Pd in applications requiring stability at high temp, such as jot anging & missile parts: Os is
Nilouluiii, Nil-43, Duthonium Du $11^{\circ}$	the densest of all elements & used for production of very hard allows for fourtain non-ting
Kullellulli, KU-44 &	increment nivers models & cleatric contents Dt & Dt set 1 at the large factor of the
$\frac{11}{2}$	instrument pivots, needle & electric contacts: Pt & Kn catalysts in the manufacture of nitic
Gold, Au-79	acid, a key ingredient in tertilizer; Ir to make high-purity crystals that have applications in
Silver, Ag-47	medical, petroleum & security industries; Ir& Ru catalysts in making acetic acid.

	Au: for jewelry; in electronics & computers, medicine & dentistry, and medals & statues.
	Ag: In electronics, coins, jewelry and medicine.
4. Nuclear Metals	U: As fuel for nuclear fission reactors; in defense to power nuclear submarines and in nuclear
Uranium, U-92	weapons.
Thorium, Th-90	Th: As fuel for Th-based nuclear reactors like breeder & MSR (molten salt reactors), which
Zirconium, Zr-40	are under development.
Hafnium, Hf-72	Zr: In high-temperature ceramic industries, Hf-free Zr as cladding material; in alloys,
Beryllium, Be-4	chemical industry, refractories (steel and glass works), medicine, tanning and oil industries.
	Hf: For nuclear control rods, alloys and high-temperature ceramics.
	Be: As metal: as an alloying agent in aerospace and defense industries; X-ray window, as
	canning material in nuclear reactors; for high-speed computers and audio components As
	oxide: electrical insulator, in microwave communications, alloys (Be-Cu in electrical and
	electronic industries; Be-Al as hardener in Al-Mg alloy melting).
5. Alloy Metals	Ti: As metal: as an alloying agent in aerospace and defense industries; X-ray window, as
Titanium, Ti-22	canning material in nuclear reactors; for high-speed computers and audio components As
Vanadium, V-23	oxide: electrical insulator, in microwave communications, alloys (Be-Cu in electrical and
Chromium, Cr-24	electronic industries; Be-Al as hardener in Al-Mg alloy melting).
Nickel, Ni-23	V: For Ti-, Ferrovanadium- and V-steel alloys; minor alloy metal which toughens steel; by
Niobium, Nb-41	adding V to any steel, helps remove oxygen and nitrogen, and gives uniform grain size;
Tantalum, Ta-73	V2O5 is used as a mordant, a material that permanently fixes dyes to fabrics.
Tin, S <mark>n-50</mark>	Cr: Primarily in stainless steel and super-alloys.
	Nickel: For alloys, batteries, coins, cars, mobile phones, jet engines, cutlery and bathroom
	taps and shower heads.
	Nb: In making steel; Zr-2.5% Nb alloy for pressure tubes in heavy water nuclear reactor, Zr-
	Nb-Cu for garter springs, SS super-alloys, superconductors, micro-alloyed steel; magnetic
	films of Fe-Nb nitride used in corrosion resistance.
	Ta: In electric components, most capacitors; transmitting and vacuum tubes, heating
	elements and heat shield; carbides for tools, magnetic films of Fe-Ta nitride used in corrosion
	resistance
	Sn: As protective coatings and alloys such as soft solder, pewter, bronze, phosphor bronze
	& steel; Nb-Sn alloy for superconducting magnets.
6. Toxic Metals	Lead: For batteries, cable sheaths, machinery manufacturing, shipbuilding, light industry,
Lead, Pb-82	lead oxide, radiation protection.
Arsenic, As-33	Arsenic: Lumber preservatives, pesticides and semiconductors.
7. Fertilizer	Phosphate: Major amount for the manufacture of phosphate-fertilizers and feed. Minor
Commodities	amount for industrial phosphates that have broad range of end-uses such as food, detergents,
Phosphate	personal hygiene and construction.

(\*Source: (1) REEs: Singh, Yamuna, 2020, Table 1.6, pp. 20-22; and Rest (2 to 7): Dhana Raju, 2020, Table 2, p. 924). Coordinates of the places, in the alphabetical order, cited in the text and Tables – 2 to 5: Ajmer-

26.4499°N,74.6399°E; Allapatna-12.3194°N,76.755°E; Bastar-19.1071°N,81.9535°E; Bhatin-

22.6643°N,86.3250°E; Boleng-28.3360°N,94.9617°E; Challanpara-8.88483°N,81.5661°E; Chavara-8.9908°N,76.5404° E; Chhatrapur-19.3597°N,84.9887°E; Diguvapalle-14.8586°N,77.9625°E; Domiasiat-25.3367°N,91.2258°E; Giddankip alle-4.3122°N,78.3572°E; Gogi-16.7321°N,76.7481°E; Gomaghat-25.21267°N,91.46732°E; Jahaz-27.7377°N,75.6343° E; Kalingapatnam-18.3387°N,84.1211°E; Keruadungri-

22.213207°N,85.876770°E; Kollam-8.9969°N,76.8721°E; Kotalu-4.3353°N,78.225°E; Malkangiri-18.3436°N:81.8825 °E; Marlagalla-2.3031°N,76.755°E; Mewar-25.25°N,74.75°E; Narwapahar-22.6962°N,86.2584°E; Nimidih-22.9917°N, 86.1459°E; Pdengshakap-25.2191°N,92.1504°E; Rachakuntapalle-4.3075°N,78.2967°E; Rohil-27.5569°N,75.49028°E; Surda-22.5540°N,86.4466°E; Tal-25.6238°N,74.0564°E; Teri sand complex-8.00°-9.50°N,77.30°-

79°E; Tummalapalle-14.3214°N,78.2678°E; Ukinal-16.7575°N, 76.6635°E, Vattalakki-11.1283°N,76.7253°E and Wahkyn-25.3214°N,91.0833°E.

#### INDIAN EXAMPLES OF CRITICAL MINERALS, RECOVERABLE FROM RADIOACTIVE MATERIALS

There are numerous Indian examples of critical minerals (CMs), recoverable from diverse radioactive materials (RMs). Some of these on which the author has the information and work-experience are presented, together with the contents of such CMs in RMs and a brief geological account of each occurrence, in the following, which by no means is exhaustive and may be taken as a guide for more comprehensive future work by interested young geoscientists.

# 1. Rare Earth Elements (REEs) and Associated other CMs in Radioactive Materials (RMs)

Of the REEs, the Light REEs have been recorded (i) mostly in the mineral (M), Th-bearing monazite in the coastal heavy mineral sand (HMS) deposits along both the eastern and western coasts and less in the inland HMS deposits in the states of Odisha, Andhra Pradesh, Tamil Nadu and Kerala; (ii) in the uraniferous sandstones (whole-rock, WR and bromo-heavies, Br.H.) of the U-prospects of Pdengshakap (for the coordinates of places cited in the text and Tables -2 to 5 see at the end of Table 1) and Gomaghat and U-deposit at Domiasiat, Meghalaya, (iii) in the fossil-placer (WR) from Boleng, Arunachal Pradesh; (iv) in the collophane (M) from the phosphatic limestone from Ukinal, Karnataka; and (v) in the U-minerals of the U-ore (O, Uraniferous Phosphatic Siliceous Dolostone, UPSD) of the U-deposit at Tummalapalle, Andhra Pradesh; and the Heavy REEs from (vii) the xenotime (M) of inland RE-placers in the states of Chhattisgarh and Jharkhand;

(viii) the garnet (M) in the coastal HMS deposits and Teri inland sand complex, Tamil Nadu and (ix) Uraninite-concentrates (MC) from the U-deposits of Narwapahar, Bhatin, Keruadungri and Surda in the Singhbhum Shear Zone (SSZ), Jharkhand (Table 2).

An examination of Table 2 points out the following:

- (i) The CMs like REEs, U-Th, Pb, Zr, Ti, As and P are in notable concentrations in diverse radioactive materials (RMs), such as whole-rock (WR), ore (O), mineral (M), heavy mineral sands (HMS) and mineral concentrates (MC), and could be recovered by adopting appropriate techniques of ore dressing and extractive metallurgy.
- (ii) For the extraction of REEs, the two minerals, namely monazite and xenotime, are the wellknown candidates, respectively, for LREEs and HREEs, besides bastnasite for LREEs which, however, is rare in India. Of these, monazite is a low-content (usually < 1 wt. % in raw sand), ubiquitous heavy mineral, mostly in the coastal and less in the inland placer heavy mineral sand (HMS) deposits, both along the east and west coasts of India, and their offshore regions. Detailed information on the placer HMS deposits of India can be had from the papers published in the edit-volumes of Dhana Raju et al. (2001) and Dhana Raju (2021). The Indian resource of monazite is 12.47 Mt (IBM yr. book, 2019). As monazite contains > 50% LREEs (Table 2, item no. 1), a resource of nearly 7 Mt of LREEs is, thus, available for their recovery and extraction of individual LREEs. JOJZ

 Table 2. REEs and Their Associated Other CMs, recoverable from diverse types of Radioactive Materials (RMs)

Tuble 2. KEEs and Their Associated Other CMs, recoverable from aiverse types of Radiouctive Materials (RMs)		
<b>REEs and Their Associated Other CMs in diverse types of RMs</b>	Reference	
A. Light REEs: 1. Monazite (M) from coastal (i, ii and iii) & inland (iv) placer HMS deposits:	Mohanty et al.,	
(i) Chhatrapur, Odisha: (wt. %, av., n=10): La2O3-11.66, Ce2O3-27.46, Pr2O3-3.23, Nd2O3-11.25,	2003, Table 1.	
Sm2O3-1.97, Eu2O3-0.08, Gd2O3-1.08, Y2O3-0.48, ThO2-10.18, UO3-0.26 & PbO-0.31.(ii)	Panda et al.,	
Kalingapatnam, Andhra Pradesh: (wt. %, av., n=6): La2O3-11.39, Ce2O3-27.17, Pr2O3-2.73,	2003, p. 434.	
Nd2O3-10.32, Sm2O3-1.58, Gd2O3-0.9, Dy2O3-0.03, Er2O3-0.07, Y2O3-0.21, ThO2-14.72 and	Krishnan et al.	
UO2-0.29. (iii) Chavara, Kerala: (wt. %) La-10.3, Ce-17.4, Nd-9.4, Sm-1.4, Eu-0.012, Tb-0.04,	2001, p.12 9.	
Yb-0.004, Th-5.80, U-0.32 and Sc (ppm)-4,500. (iv) Siri river, Chhattisgarh: (wt.%, av., n=15)	Sarbajna et al.,	
La2O3-12.2, Ce2O3-26.6, Pr2O3-2.7, Nd2O3-8.8, Sm2O3-1.6, Gd2O3-1.3, Dy2O3-0.5, Er2O3-	2007	
0.2, ThO2-11.3, UO2-0.5		
2. Lower Mahadek Uraniferous Sandstone, Meghalaya (WR in ppm; Br. Heavies. in wt. %):	Dhana Raju et	
(i) Pdengshakap: WR: La-210, Ce-368, Nd-NA, Sm-7.8, Eu-1.7, Tb-2.0, Yb-4.5, Lu-0.26, Y-58.	al., 1989,	
BrH: La-2.148, Ce-3.513,Nd-1.51,Sm-0.305,Eu-0.0576,Tb-0.028,Yb-0.077,Lu-0.013,Y-1.07.	Tables 7,6	
(ii) Gomaghat: WR: La-92, Ce-137, Nd-33, Sm-11, Eu-1.20, Tb-2.20, Yb-37, Lu-1.70, Y-NA.	(Sample Nos.	
Br.H: La-0.030,Ce-0.05,Nd-ND,Sm-0.005,Eu-0.0002,Tb-0.001,Yb-0.045,Lu-0.001,Y-0.043.	916, 927, 934)	
(iii) Domiasiat: WR: La-224, Ce-381, Nd-165, Sm-24, Eu-1, Tb-2, Yb-2.8, Lu-0.35, Y-NA.		
BrH: La-1.55, Ce-2.90, Nd-1.05, Sm-0.19, Eu-0.003, Tb-0.014, Yb-0.012, Lu-0.002, Y-0.21.		
NA: Not analysed; ND: Not determined. Analytical Method: INAA.		

3. Fossil-placer (cyrtolite-Ti ht. ironstone, WR), Boleng, E. Siang dist., Arunachal Pradesh:	Singh et al,
(In wt. %, av., n=3), (RE, Y)2O3:0.58-3.1, ZrO2:1.66-5.4, ThO2:0.13-0.5, U3O8:0.02-0.087,	1994,
TiO2:13.3-15.0, Fe2O3:41.84-60.31, FeO: 1.33-1.6. Sp. Gr.: 3.96-4.66; 60-84% heavy min.	p. 410.
4. Collophane (M) from (a) Phosphatic Limestone, Ukinal, Karnataka:	Dhana Raju
(In wt. %, av., n=4) La2O3-0.02, Ce2O3-0.03, Pr2O3-0.06, Nd2O3-0.01, Sm2O3-0.10, Gd2O3-	et al., 2002
0.06, Yb2O3-0.05, UO2-0.16, ThO2-0.07, PbO-0.06 and P2O5-40.61.	p.313.
5. U-phases in the UPSD (Ore), Tummalapalle, Andhra Pradesh (in wt. %, av., n: no. of spots)	Roy and Dhana
(a) Pitchblende (n:8): Ce2O3-0.33,Gd2O3-0.05,Dy2O3-0.06,Y2O3-0.07,V2O5-0.28,As2O3-0.30	Raju, 1999, p.
(b) Coffinite (n:18): Ce2O3-0.40,Gd2O3-0.11,Dy2O3-0.10,Y2O3-0.29,V2O5-0.41,As2O3-0.28	69.
(c) U-Si Complex(n:6) Ce2O3-0.45,Gd2O3-0.10,Dy2O3-0.08,Y2O3-0.24,V2O5-0.28,As2O3-0.08	
	Madhuparna et
6. REE-Th-Zr-Ti-rich Tremolite-Actinolite rock, Vattalakki, Kerala:	al., 1990, p.
(WR, in ppm, av., n:5): La-9022, Ce-14780, Pr-1972, Nd-5707, Sm-345, Eu-15, Gd-150, Tb-51,	138,136.
Dy-105,Yb-8.4,Y-245,Cr-571,Sc-29,Co-46 and Ce/Yb:1863; TiO2-8.86%, ThO2-0.88%.	
	Sarbajna et al.,
7. Heavy REEs: 7. Xenotime (M) from Siri River RE placers, Chhattisgharh (in wt. %, av.)	(2007)
(n:25) Y2O3-43.9, La2O3-0.0, Ce2O3-0.1, Pr2O3-0.1, Nd2O3-0.3, Sm2O3-0.1, Gd2O3-3.1,	
Dy2O3-4.4, Er2O3-4.6, Yb <mark>2O3-5.4, ThO2-0.6</mark> UO2-1.6, PbO-0.6 and P2O5-33.4.	Panda et al.,
8. Garnet (M) from Kalingapatnam beach placers, Andhra Pradesh (in wt. %, av., n: no. of points)	2017, p.133
(n: 5): Y2O3-0.04, La2O3-0.01, Ce2O3-0.02, Nd2O3-0.03, Gd2O3-0.05, Dy2O3-0.06, Er2O3-0.10,	
Lu2O3-0.11, ΣLREEO-0.07, ΣHREEO-0.32 and ΣREEO-0.39.	Rao & Rao,
9. Uraninite-concentrates (MC) from U-deposits* in the Singhbhum shear zone, Jharkhand.	1980, p.
(In wt. %, n: 5, range): UO2:40.02 to 51.14, UO3: 22.95 to 31.18, ThO2: 0.18 to 0.84, ΣRE2O3:	388.
4.59 to 8.68, with predominance of heavy (Gd, Dy & Er) over light lanthanides.	
*Narwapahar, Bhatin, Keruadungri and Surda.	

- (iii) Of the LREEs in the Lower Mahadek sandstone, the whole-rock samples contain REEs in ppm level with the highest content of Ce being 381 ppm, whereas their bromoform-heavies (Br. H.) contain in %-level, with the highest content of Ce being 3.513% (Table 2, item no. 2) in the Domiasiat Uore; the % Br. H. in the whole-rock sandstone is usually < 1% with a maximum of 4.9%. This pattern demonstrates that for recovering the REEs in notable content from low-content whole-rocks, physical beneficiation, involving separation and enrichment of REE-bearing heavy minerals like monazite and zircon using the gravity, electrical and electro-magnetic methods should be the first step in mineral progressing. Detailed account of these sandstones from the U-prospects of Pdengshaka and Gomaghat, and the sandstonetype U-deposit of Domiasiat can be had from Dhana Raju et al., (1989), Sen et al., (2002) and Dhana Raju (2019).
- (iv) The cyrtolite-Ti-hematite ironstone (Table 2-item no. 3) from Boleng, Arunachal Pradesh is a rare fossil-placer (Sp. Gr. 3.96-4.66) containing 60-85% heavy and resistant minerals comprising 46-61% opaque minerals (mostly Ti-hematite), 9-13 rutile, 3-9% cyrtolite, 0.7-4.5% monazite and 0.2-0.9% thorite, all set in 10-30% sericitic matrix.

With such dominant heavy mineral assemblage, it is no surprise that it contains notable contents of critical minerals of Ti, Zr and REEs in major amounts. Physical beneficiation of this rock will lead to much higher concentrations of these critical minerals. In view of the importance of this rock for the above critical minerals, it is suggested that a detailed exploration should be carried out in the area for locating more bodies of this rock. The rock is recorded in association with the quartzite and metagraywacke of the Proterozoic Miri Group (Singh et al., 1994).

- (v) The critical mineral (P, LREEs, U-Th-Pb) data of collophane from the Ukinal area in Karnataka (Table 1-item 4) points out that this rock, near to the Gogi high-grade U-deposit, needs to be reexplored to evaluate the resource of critical minerals listed above.
- (vi) Data on the critical minerals (REEs, V and As) in the U-phases of the rare carbonate-hosted Udeposit at Tummalapalle (Table 2 - item 5) highlight the importance of extracting them as byproducts of U during the ongoing mineral processing of this deposit, which will be an important value-addition.
- (vii) The REE-Th-Zr-Ti-rich tremolite-actinolite rock from Vattalakki, Kerala (Table 2 -item 6) is another rare rock that is important for the above

critical minerals. As this rock occurs as a sill-like intrusive body in the gneisses in the area, in association with dolerite sill, it is recommended that a detailed exploration in the area should be undertaken for such exotic rocks to enhance the resource position of important critical minerals, listed above.

- (viii) As xenotime is the HREE-mineral with notable contents of CMs of individual HREEs, together with other CMs like U-Th, Pb and P (Table 2 item 7), AMD has been exploring and exploiting since long the potential inland riverine placer bodies for both xenotime and monazite in the states of Chhattisgarh and Jharkhand, with their source rocks being granites and granite pegmatites.
- (ix) An HM concentrate comprising 2-5% xenotime, 22-54% monazite, 45-50% Fe-minerals, 5-24% garnet, 2-6% zircon and 12% other minerals, and analyzing 1.5-3.5% Y2O3 and 2.0-12.81% Ce2O3 has been recovered by mobile pilot-scale recovery plants (Ramesh Babu, 2021). More details on the Indian RE-inland placer deposits, exploration, exploitation, resources etc., can be had from the papers in (Eds.) Mahadevan and Dhana Raju (1999), Singh (2020) and (Ed.) Dhana Raju (2021).
- (x) Garnet (Table 2 item 8), a major heavy mineral in the coastal and some inland (like Teri HMS deposits in Tamil Nadu) is a known host for REEs though in minor content ( $\sum$ REEO ~ 0.3-0.4%), with HREE/LREE being 4-5. As the garnet resource in the Indian placers is 187 Mt (IBM Indian Mineral Year Book 2018/2019), taking its  $\sum$ REEO around 0.3%, an REE resource of 0.56 Mt is available for recovery and extraction, with 5/6th of it being HREEs. Hence, recovery of HREEs from more available garnet than xenotime should also to be taken up, along with the recovery from xenotime.
- (xi) The uraninite-concentrate from different Udeposits of the Singhbhum shear zone (SSZ), Jharkhand (Table 2- item 9) contains notable content of ∑RE2O3: 4.59 to 8.68, with predominance of heavy (Gd, Dy & Er) over light lanthanides, and, hence, extraction of REEs as byproducts from uraninite during the processing of U-ore should be taken up under value-addition so as to improve the resource base particularly of HREEs. At the Jaduguda U-mine in SSZ, a byproduct recovery plant was in operation for recovering Cu, Ni, Mo and Fe (as magnetite) from U-ore (IAEA, 2001).

# 2. Battery Metals and Precious Metals as CMs in Radioactive Materials (RMs)

Of the three battery metals that can be recovered from radioactive materials (RMs), Li occurs in the form of its discrete minerals, viz., spodumene, lepidolite and amblygonite in the granite-related rare metal pegmatites of Marlagalla, Karnataka (Krishna and Thirupathi, 1999) and Bastar – Malkangiri Pegmatite Belt (BMPB) in Chhattisgarh and Odisha (Ramesh Babu, 1999; 2021); and Co and Ni occur in the sulphides of in the U-ore of

Gogi, Karnataka (Dhana Raju et al., 2002). Among the precious metals, PGMs at ppb level occur in the rare REE-Th-Zr-Ti-rich tremolite-actinolite rock in the Vattalakki area, Kerala (Dhana Raju et al., 1998), whereas Au and Ag at per cent level occur in the sulphides of the U-ores at Gogi, Karnataka (Dhana Raju et al., 2002) and Tummalapalle (Roy and Dhana Raju, 1999) (Table 3).

An examination of the data on the three battery CMs (Table 3) points out the following:

- (i) Li occurs in its discrete minerals such as spodumene, lepidolite and amblygonite, which are hosted usually by the granite-related rare metal pegmatites that occur as well-defined belts, groups and clusters in the states of Chhattisgarh, Odisha and Karnataka.
- (ii) AMD has been extensively exploring and exploiting these pegmatites with recovery of spodumene mostly from the rare metal pegmatites in the Marlagalla area, Karnataka, and lepidolite and amblygonite from those in the states of Chhattisgarh and Odisha by field-based mobile plants (Krishna and Thirupathi, 1999; Dwivedy, 1996; Ramesh Babu, 1999 and 2021).
- (iii) Of the above three Li-minerals, the Li-content ranges from 4.6 to 9.1%, with the lowest, intermediate and highest contents, respectively, in lepidolite, spodumene and amblygonite.
- (iv) The other two battery CMs, namely Co and Ni occur in recoverable amounts in the sulphides of the U-ore (hosted both by limestone and granite) at Gogi, Karnataka. These sulphides include pyrite (euhedral and framboidal), chalcopyrite and galena.
- (v) Of these three sulphides, chalcopyrite and pyrite host higher contents, and galena carries low contents of Co and Ni. Between the two forms of pyrite, viz., euhedral and framboidal, the euhedral one in granite and the framboidal one in limestone carry relatively higher contents of both Co and Ni (Dhana Raju et al., 2002).

(vi) These patterns indicate that both the pyrite and chalcopyrite should be separated and concentrated from the Gogi U-ore by appropriate mineral processing and extraction techniques to recover Co and Ni as high-value by-products required for the battery industry.

Table 3. Battery Metals and Precious Metal.	s, recoverable from Radioactive Materials (I	RMs)
---	--	------

Battery Metals and Precious Metals as CMs in RMs	Reference
A. Battery Metals	
1. Lithium (Li): (in wt. %)	Krishna & Thiru-
(i) Marlagalla, Karnataka: Li2O: (M) Spodumene: 6.55-7.35, av. 7.14 (n=4),	pathi,1999,p. 146
(ii) Govindpal, BMPB, Chhattishgarh & Odisha: Lepidolite: 4.55, Amblygonite: 9.12	Dwivedy, 1996, p.
	113, 114
2. Cobalt (Co) & Nickel (Ni): (a) Sulphides in Gogi U-ore: (in wt. %, av., n=6-45), Pyrite	
(euhedral): (i) in L.St.: Co-0.04, Ni-0.04, (ii) (in Gr.): Co-0.10, Ni-0.06	Dhana Raju et al.,
Pyrite (framboidal): (i) in L.St.: Co-0.36, Ni-0.14, (ii) (in Gr.): Co:0.03, Ni:0.0	2002, p.312
Galena (i) in L.St.: Co-0.02, Ni-0.01, (ii) (in Gr.): Co-0.01, Ni-0.01	
Chalcopyrite (i) (in L.St.): Co-0.28, Ni-0.18 (L.St: Limestone, Gr.: Granite)	
(b) Sulphides in Tummalapalle U-ore: (in wt. %, av.), Pyrite (euhedral): Co-0.02,	Roy & Dhana Raju,
Ni-0.07 (n=19), Pyrite (framboidal): Co-0.03, Ni-0.085 (n=14)	p.70
B. Precious Metals	
3. Platinum Group Metals (PGMs) (in ppb, av., n=4), REE-Th-Zr-Ti-rich tremolite-actinolite	Dhana Raju et al.,
rock, Vattalakki, Kerala: Pt: 3.125, Ir: <0.5-0.5, Rh: <0.5, Ru:1.0, Pd-2.25	1998 p. 525
4. Gold (Au) in sulphides of the U-ore at Gogi, Karnataka (in wt. %, av., n=6 to 45)	Dhana Raju et al.,
Pyrite (euhedral): (i) in L.St.: Au-0.00, (ii) (in Gr.): Au-0.04	2002, p.312
Pyrite (framboidal): (i) in L.St.: Au-0.00, (ii) (in Gr.): Au:0.09	
Galena (i) in L.St.: Au-0.04, (ii) (in Gr.): Au-0.07	
Chalcopyrite (i) (in L.St.): Au-0.00, (L.St: Limestone, Gr.: Granite)	
5. Silver (Ag) in sulphides of the (a) Gogi U-ore, Karnataka (in wt. %, av., n=6 to 45)	Dhana Raju et al.,
Pyrite (euhedral): (i) in L.St.: Ag-0.04, (ii) (in Gr.): Ag-0.02; Pyrite (framboidal): (i) in L.St.:	2002, p.312
Ag-0.09, (ii) (in Gr.): Ag:0.04; Galena (i) in L.St.: Ag-0.00, (ii) (in Gr.): Ag-0.02; Chalcopyrite	CO70
(i) (in L.St.): Ag-0.26 (L.St: Limestone, Gr.: Granite)	Roy & Dhana Raju,
(b) Tummalapalle U-ore, A.P.: Pyrite (euhedral): 0.04%, Pyrite (framboidal): 0.03%	1999, p. 70

- (vii) Amongst the precious CMs, as expected, the contents of Ag > Au >> PGMs, with the first two being in notable, low per cent grade in the sulphides of the Gogi U-ore (Dhana Raju et al., 2002), whereas the contents of PGMs in the rare tremolite-actinolite rock at Vattalakki, Kerala (Dhana Raju et al., 1998) appear to be of academic interest; however, detailed exploration in the area and its contiguous areas is suggested for more possible such bodies.
- (viii) Regards Au and Ag in the Gogi area, pyrite and galena in granite-hosted U-ore carry higher contents of Au than those in the limestone-hosted U-ore, whereas pyrite and chalcopyrite in the limestone-hosted U-ore carry higher contents of Ag, with chalcopyrite carrying the highest content at 0.26% Ag. These data, thus, cumulatively demonstrate the importance of separation and

All rights are reserved by UIJRT.COM.

concentration of sulphides from the Gogi U-ore to recover both precious and battery CMs as valuable by-products as and when the ore is processed for U as the main-product.

# 3. Nuclear Metals as CMs in Radioactive Materials (RMs)

Of the five nuclear metals as CMs in RMs, uranium (U) occurs in India in low-grade (< 0.2% U3O8) deposits, with source of U mostly in granitic rocks and hosted by diverse rock types such as (i) granite and limestone (hydrothermal type) in the Neoproterozoic Bhima basin in the Gogi area, Karnataka, (ii) impure carbonate-sediment (UPSD – uraniferous, phosphatic, siliceous dolostone – in the Tummalapalle - Rachakuntapalle - Giddankipalle sector (and its contiguous areas (rare strata-bound, syn/dia-genetic type with a large-tonnage

[> 1.5 lakh tonnes U3O8, accounting for > 50% of Indian U-resources of > 3 lakh tonnes U3O8]) along the SW, S and W margins of the Mesoproterozoic Cuddapah basin in Andhra Pradesh, (iii) basement granite and its overlying quartzite of the unconformityproximal type along N- and NE-margins of the Cuddapah basin in parts of Telangana and Andhra Pradesh; (iv) diverse low-medium grade tourmalineapatite-magnetite-biotite-chlorite-quartz schists, quartzite, sheared conglomerate etc., along the Singhbhum Shear Zone (SSZ) in Jharkhand; (v) Upper Cretaceous, Lower Mahadek sandstone in the Domiasiat area, Meghalaya; (vi) albitites and albitised associated rocks (Albitisation-related, metasomatic hydrothermal type) in Rajasthan, etc. From these U-ores, U as the main product is recovered by mineral processing either through acid-route in case of silicate rocks as host (e.g., U-ores of different deposits in SSZ, Jharkhand) or alkali-route in case of carbonate-rocks as host of Uminerals (e.g., Tummalapalle U-deposit in Andhra Pradesh). For more details, see the papers in AMD's EARFAM, v. 14, 2002, (Eds. R. Dhana Raju et al.,) and Dhana Raju, (2019). In India, most of Thorium (Th) is in the mineral, monazite (4-15% ThO2) that is a finegrained constituent (< 1% in raw sand) of the placer heavy mineral sand (HMS) deposits along the east and west coasts, and less in the inland HMS deposits in the states of Odisha, Andhra Pradesh, Tamil Nadu and Kerala (for details, see the papers in AMD's EARFAM

Spl. Vol. 13 on 'Beach and Inland Heavy Mineral Sand Deposits of India', (Eds.) Dhana Raju et al., 2001) and in the book on 'Indian Placer Deposits' (Ed.) Dhana Raju (2021). Zirconium (Zr) and hafnium (Hf) in recoverable contents occur mostly in the form of finegrained mineral, zircon in the HMS deposits cited above, and beryllium (Be) occurs in the mineral, beryl in the Rare Metal Pegmatite deposits in the states of Chattisgarh, Odisha and Karnataka, from which it can be recovered as the main product (for details see papers in the AMD's EARFAM Spl. Vol. 12 on 'Rare Metal and Rare Earth Pegmatites of India', (Eds. Mahadevan and Dhana Raju (1999) and in the book on 'Indian Placer Deposits' (Ed.) Dhana Raju, 2021) (Table 4).

An examination of the data in Table 4 points to the following:

(i) Out of different U-deposits in India, the one at Tummalapalle is the lowest (0.045% U3O8) and that at Gogi is the highest (0.18% U3O8) in the Ugrade. The U-ore in different deposits of the SSZ, Jharkhand contains many CMs, some of which like Cu, Ni and Mo have been under recovery as by-products, whereas Co, Zr and Y, and Re from the molybdenite-concentrate may be recovered by appropriate extraction technologies that may need R & D.

**Table 4.** Nuclear Metals and Their associated other CMs, recoverable from different types of Radioactive Materials(RMs)

Nuclear Metals as CMs, recoverable from RMs	Reference
1.Uranium (U)	Dhana Raju, 2019
(i) Gogi U-deposit, Karnataka: U-ore: U3O8: 0.013-1.678%, av. 0.18%	p. 126
(ii)Tummalapalle U-deposit, Andhra Pradesh: U-ore: U3O8: av. ~ 0.045%	-do-, p. 246
(iii)Unconformity-proximal type, Telangana, AP: U-ore: U3O8: 0.05-0.1%	-do-, p. 185
(iv)Jaduguda-Bhatin-Nimdih U-deposits in SSZ: U-ore: eU3O8: 0.05-0.07%; (in ppm, av. of	Dhana Raju, 2019,
359 samples) Ni-402, Cu-70, Mo-47, Co-47, Y up to 290 and Zr up to 230;	p. 68
by-product molybdenite-concentrate (UCIL-recovery plant) contains upto180 ppm Re	Singh et al., 1979
(v) Sandstone-type U-deposits in Meghalaya: U-ore: (a) Domiasiat: U3O8: 0.098%,	Dhana Raju, 2019,
(b) Wahkyn: 0.124% eU3O8	p. 395-do- , p.326
(vi) Albitisation-related type U-deposit: Rohil, Rajasthan:U-ore: U3O8: 0.05-0.08%	
2. Thorium (Th): mostly in monazite of HMS deposits: 5-15% ThO2 (see Table 2-A 1)	
	Ravi, 2021, p. 527
3. Zirconium (Zr) content (in wt.%) of Zircon from HMS deposits:	
(i) OSCOM, Chhatrapur, Odisha: 64.5, (ii) Manvalakuruchi, Tamil Nadu: 65.8,	
(iii) Chavara, Kerala: 65.3 and Kollam, Kerala: 64.8	
	Krishna &Thiru-
4. Beryllium (Be) & associated other CMs (wt.%) in Beryl from RM pegmatite deposits	pathi, 1999, p. 136
(i) Marlagalla, Karnataka: BeO-11.89, Li2O-0.55, Cs2O-0.66 and Rb-799 ppm	Ramesh Babu
(ii) Challanpara (in BMPB), Chhattisgarh: BeO-13.20, Cs2O-0.06and Rb2O-0.005	1999, p.17
(iii) Ajmer, Rajasthan: BeO-13.41, Cs2O-986 ppm and Rb2O-41 ppm	Maithani & Nagar,
	1999, p. 105

- (ii) From the mineral monazite in the HMS deposits, both LREEs and Th, which occur as major constituents, are recoverable.
- (iii) The CMs, zirconium and its ubiquitous associate, Hf in the mineral, zircon from the HMS deposits have been under recovery in the nuclear industry.
- (iv) As the CM, Be is associated with the other CMs, namely Cs and Rb in recoverable contents in the mineral, beryl of the Rare Metal pegmatites, R & D may be required to extract both Cs and Rb as byproducts, when Be is recovered as the main product from the mineral, beryl.

# 4. Alloy metals as CMs in Radioactive Materials (RMs)

Of the alloy metals, Ti, Nb-Ta and Sn occur usually in major amounts, whereas V, Cr and Ni occur in minor to trace contents in the radioactive materials (RMs). Thus, Ti is a major constituent and Cr is in minor to trace amounts (<1%) in the mineral, ilmenite (FeO.TiO2) that is normally in the maximum amount amongst the valuable heavy minerals (ilmenite-sillimanite-garnet, each > 1% and zircon-rutile-monazite, each < 1%) in the HMS deposits along the east and west coasts of India (see papers in the volumes edited by Dhana Raju et al., 2001 and Dhana Raju, 2021). Nb-Ta occur together in major amounts in some rare minerals such as columbitetantalite, microlite, ixiolite and wodginite, and Sn in major amount in the mineral, cassiterite; all these minerals are valuable constituents in the rare metal pegmatites in the BMPB in the states of Chhattisgarh and Odisha (Ramesh Babu, 1999 and 2021) and in the Marlagalla - Allapatna area in Karnataka (Krishna and Thirupathi, 1999). V, Ni, Cu and Mo in notable contents occur in the sulphides (pyrite, chalcopyrite, pyrrhotite and molybdenite) of the uraniferous albititites and related rocks in the albitisation-related U-deposits of Rohil - Jahaz in Rajasthan (Padhi et al., 2016) and Uores of Tummalapalle and Gogi, and in relatively higher contents in their sulphides (Dhana Raju, 2019) (Table 5).

 Table 5. Alloy Metals and Their associated other CMs, recoverable from different types of Radioactive Materials (RMs)

A <mark>lloy M</mark> etals as CMs, recoverable from Radioactive Materials (RMs)	Reference
1. Titanium, Chromium and Vanadium in HMS deposits	
(i) Chhatrapur HMS deposit, Odisha: Ilmenite: (wt. %, range, n=20) TiO2-45.67 to 55.41,	Rao & Sengupta,
Cr2O3-0 to 0.089, V2O5-0.247 to 0.306	2014, p. 5
(ii) HMS deposits in Andhra Pradesh: Ilmenite: (wt. %, range, n=7) TiO2-48.0 to 50.8, V2O5-	Ravi, 2021, p. 411
0.07 to 0.25; Cr (in ppm, n=3): 650-1178 (Jagannadha Rao et al., 2015)	Chandrasekaran
(iii)HMS deposits in Tamil Nadu: Ilmenite: (wt. %) (a) Coastal beach & dune sands: (range,	& Murugan, 1999,
n=68) TiO2: 49.7-57.8, Cr2O3: < 0.1 to 0.7, V2O5-0.1-0.3; (b) Inland Teri sands (n=20): TiO2-	рр. 94-95
49.7 to 54.5, Cr2O3-0 to 0.2, V2O5-0.2-0.25	Krishnan et al.,
(iv)Chavara HMS deposit, Kerala: Ilmenite: (wt. %, n=9) TiO2-58.4 to 65.0, Cr2O3-0.15-0.25,	2001, p. 128
V2O5-0.15-0.25	
2. Niobium-Tantalum in Nb-Ta Ores of Rare Metal Pegmatites	Ramesh Babu,
(i) BMPB, Chhattisgarh and Odisha: (wt. %) (a) Columbite-Tantalite: Nb2O5-35.2, 32.6; Ta2O5-	1999, p. 19
31.2, 37.1; (b) Ixiolite: Nb2O5-11.0, 8.5; Ta2O5-57.1, 60.5; (c) Niobian Ixiolite: Nb2O5-58.8,	
Ta2O5-14.2; (d) Wodginite: Nb2O5-2.0, 32.6; Ta2O5-67.8; (e) Microlite: Nb2O5-3.8,	
Ta2O5-69.8	Sarbajna et al.,
(ii)Marlagalla-Allapatna, Karnataka: (wt. %) (a) Columbite-Tantalite (n=8): Nb2O5:18.66-	1999
68.67,Ta2O5:16.34-59.56;(b) Microlite: Nb2O5- 4.04, 4.31; Ta2O5-73.40, 73.01	
3. Tin in Sn-ore of Rare Metal Pegmatites	
(i) BMPB, Chhattisgarh and Odisha: Cassiterite: (wt. %, range, n=4) SnO2-79.62 to 91.40,	Dwivedy, 1996, p.
Nb2O5- 0.78 to 2.60, Ta2O5-4.01 to 12.40, WO3-<0.01 to 0.10	116
4. Nickel, Vanadium and their other associated CMs in U-ores/phases and sulphides	
(i) Albitites-microclinites-pyroxenites from Mewar-Gujarwas to Tal, Rajasthan	Singh et al., 1998,
U-Ore (n=14, ppm): U- 116 to 13,600, Ni- 22 to 180, Cu- <5 to 2,165, Mo- < 5 to >1,000,	p.5
(ii)Tummalapalle U-ore, Andhra Pradesh: (wt. %) V2O5 (av.) in U-phases: Pitchblende-0.28	
(n=8), Coffinite-0.41 (n=18), U-Si Complex-0.28 (n=6) and U-Si-Ti Complex-0.35 (n=6)	Roy & Dhana
(iii)Gogi U-ore, Karnataka: (wt. %) (a) Ni (range, av.): pyrite: 0.0 to 1.47, 0.06 (n=76), Galena:	Raju, 1999, p. 69
0.0 to 0.07, 0.01 (n=19), Chalcopyrite: 0.03 to 0.28, 0,12 (n=6);	

 (b) Cu (range, av.): Chalcopyrite: 28.59-34.12, 31.16 (n=6); pyrite: 0.0 to 0.94, 0.04 (n=76);
 Dha

 Galena: 0.0 to 0.32, 0.07, n=19)
 2002

Dhana Raju et al., 2002, p. 312

An examination of the data in Table 5 points to the following:

- (i) Of the ilmenite in the HMS deposits of India, TiO2 content is the highest (65%) in the Chavara deposit and lowest (45.67%) in the Chhatrapur deposit, whereas Cr2O3 is the highest (0.7%) in the Tamil Nadu deposits and lowest in the Chhatrapur deposit (0-0.09%) and V2O5 is the highest in the Chhatrapur deposit (0.25-0.31%) and comparable  $(\sim 0.1 \text{ to } 0.25\%)$  in the deposits of Andhra Pradesh, Tamil Nadu and Kerala. As the India's ilmenite resource is 630 Mt (IBM yr. book, 2019), recovery of Cr and V from ilmenite will add substantial amounts to their resource base. Furthermore, as the V2O5 content in the U-phases in the Tummalapalle U-ore is appreciable (0.28 to 0.41%), it should be recovered as a by-product for value-addition during the processing of this U-ore.
- (ii) In the columbite-tantalite, the ones from Marlagalla deposit contain both Nb and Ta in higher amounts as compared to that in the BMPB deposits. Besides, Ta > Nb occurs in the minerals like ixiolite, wodginite and microlite, and the inverse pattern (Nb > Ta) in the niobian-ixiolite in the BMPB.
- (iii) Cassiterite in the BMPB contains notable amounts of both Nb and Ta, which may be recovered as valuable by-products during the processing of cassiterite for tin as the main product.
- (iv) Ni is in recoverable amounts from the sulphides such as pyrite and chalcopyrite of the Gogi U-ore, whereas from the sulphides in the U-ores hosted by the albitite-pyroxenite-microclinite in Rajasthan, Cu and Mo may be recovered as mainproducts and Ni as a by-product.

# 5. Toxic Metals and Fertilizer Commodity as CMs in Radioactive Materials (RMs)

The toxic metals – lead (Pb) and arsenic (As) – occur in notable concentration in different RMs, with radiogenic Pb in all U-phases in genetically diverse U-ores and As in sulphides in these ores, as demonstrated by the following examples:

(i) In the stratabound, syn/dia-genetic impure dolostone-hosted Tummalapalle U-ore (UPSD), Andhra Pradesh, the average contents (in wt. %) of radiogenic Pb in different U-phases are as follows: pitchblende: 4.82 (n=8), coffinite: 2.48 (n=18), U-

Si complex: 1.75 (n=6) and U-Si-Ti complex: 5.58 (n=6) (Roy and Dhana Raju, 1999, p. 69);

- (ii) In the hydrothermal type Gogi U-ore, hosted both by basement granite and its overlying limestone, the average contents (in wt. %) of radiogenic Pb in different U-phases are as follows: pitchblende: 6.28 (n=9) in limestone and 2.48 (n=14) in granite; coffinite: 3.20 (n=16) in limestone and 1.27 (n=20)in granite; and U-Ti complex in granite: 0.04 (n=6) (Dhana Raju et al., 2002, p. 310). In the uraniniteconcentrates from hydrothermal type of different U-deposits in the SSZ, Jharkhand, PbO ranges from 6.29% (Surda) to 13.64% (Narwapahar) (Rao and Rao, 1980, p. 388). Besides radiogenic lead, non-radiogenic Pb occurs as the major constituent in the sulphide, galena, e.g., in the Gogi U-ore, av. 83.74% (n=6) in the limestone and av. 85.59% (n=13) in the granite.
- (iii) Arsenic (As) in lower concentration occurs in the sulphides of U-ores, e.g., in the Gogi U-ore, framboidal pyrite in both limestone and granite, respectively, contains average contents of 0.28% (n=7) and 0.17%, whereas the euhedral pyrite in limestone and granite, respectively, contains average contents of 0.23% (n=45) and 0.04% (n=7) (Dhana Raju et al., 2002, p. 312).
- The non-metal CM, phosphate used as a fertilizer (iv) commodity occurs in notable content in the phosphatic RMs containing major amount of a phosphatic mineral like collophane and apatite. e.g., UPSD in the different U-deposits of Diguvapalle, Tummalapalle, Gidankipalle and Kotalu along the SW-margin of the Cuddapah basin in Andhra Pradesh contains usually 1-9.4% P2O5 with a maximum of 28.58% P2O5 in that from Kotalu (Roy and Dhana Raju, 2012, p. 183) and, hence, the phosphate in the large tonnage (> 1.5 lakh tonnes) dolostone-hosted U-ore should be recovered as a by-product for agricultural use during the processing of this ore in the Tummalapalle plant. Another example of notable phosphate content in RA material is the radioactive limestone from the Ukinal area (near the Gogi U-deposit), Karnataka that contains 25.64% P2O5 (average, n=15) (Dhana Raju et al., 2002, p. 302).

#### DISCUSSION

### Recovery of Critical Minerals (CMs) from Radioactive Materials (RMs)

An examination of the data (Tables – 2 to 5) and information, presented in the earlier sections, points out that out of the 33 Critical Minerals (CMs) identified from the Indian perspective (Dhana Raju, 2020) plus phosphate as one more CM, > 60%, viz., As, Be, Cs, Cr, Co, Au, Hf, Li, Ni, Nb-Ta, REEs, Sc, Rb, Ag, Sn, Ti, W, U, V and Zr metals, and non-metal phosphate are in contents recoverable (mostly as by-products under value-addition and a few as main products) from diverse radioactive materials (RMs) that include whole-rock and its constituent heavy minerals, U-ores and their constituent U-phases and sulphides, rare metal minerals, mineral concentrates and heavy mineral sands. This is illustrated by the following examples:

- (i) The LREEs in the whole-rock Lower Mahadek sandstones in Meghalaya are in ppm level, whereas they are in per cent level in their bromoform-heavies (Br. H.) (Table 2 - item 2), which are usually < 1% in their host sandstone. This pattern points out that when the CMs like LREEs in a rock are in ppm level, it is advisable to identify first the minerals in the rock which account for such CMs, followed by their physical separation so as to enhance the concentration of these CMs for their better recovery as by-products as value-addition during the processing of the whole-rock. Furthermore, in rare rocks like the fossil-placer from Boleng, Arunachal Pradesh and tremolite-actinolite rock from Vattalakki, Kerala (Table 2 - items 3 and 6, respectively), ∑REEs and Ti are in recoverable, major amounts.
- (ii) All the diverse types of U-ores, such as hydrothermal, stratabound syn/dia-genetic, unconformity-proximal, albitisation-related, sandstone etc., contain some CMs (other than U and/or Th [radio-element CMs] and radiogenic Pb [toxic CM, section 5, item nos. i and ii]) in lower concentration that increases substantially in their constituent mineral phases like U-minerals and sulphides. This pattern is illustrated by the following examples of the U-ores of Gogi (Karnataka) and Tummalapalle (Andhra Pradesh): (a) In the Tummalapalle U-ore, its U-phases contain per cent level REEs, V and As (Table 2 item 5) and its sulphides contain recoverable grade (% level) Ag (Table 3 - item 5) and Co and Ni (Table 3 - item 2). (b) In the Gogi U-ore, its sulphides contain per cent level Co & Ni, Au and

Ag (Table 3 - items 2, 4 and 5, respectively) and arsenic (As, section 5 - item iii).

- (iii) The uraninite-concentrates from different Udeposits in the Singhbhum shear zone, Jharkhand, contain per cent level  $\sum$ REEs, with HREE >> LREE (Table 2 - item 9).
- (iv) The rare metal minerals in the pegmatites of BMPB in the states of Chhattisgarh and Odisha, and Marlagalla-Allapatna in the state of Karnataka contain the CMs of Li, Cs, Nb-Ta, Be and Sn in recoverable, major amounts from their respective minerals (spodumene, lepidolite and amblygonite for Li; pollucite for Cs; columbite-tantalite, ixiolite, niobian ixiolite, microlite and wodginite for Nb-Ta; beryl for Be and cassiterite for Sn), besides other CMs in minor (per cent level) contents, e.g., cassiterite containing Nb-Ta and W (Table 5 - items 2 and 3) and beryl containing Li, Cs and Rb (Table 4 - item 4).
- (v) The coastal and inland heavy mineral sand (HMS) deposits along the east and west coasts of India in the states of Odisha, Andhra Pradesh, Tamil Nadu and Kerala contain some CMs in major amounts in their respective minerals (e.g., Ti in ilmenite [Table 5 item 1], Zr in zircon [Table 4 item 3] and LREEs in monazite [Table 2 item 1]), and a few other CMs in minor amounts (e.g., HREE in garnet [Table 2 item 3]; Cr and V in ilmenite [Table 2 item 1]). Furthermore, the inland riverine placer bodies in the states of Chhattisgarh and Jharkhand contain both xenotime and monazite, from which HREEs and LREEs, respectively, can be recovered (Table 2 items 1[iv] and 7).
- (vi) Non-metal CM, Phosphate: This fertilizer commodity is in major amounts in the Tummalapalle U-ore (Andhra Pradesh) and the Ukinal U-ore (Karnataka) (section 5 item iv) and during the processing of such ores, phosphate can be recovered as a co-product of uranium. Furthermore, as the radioactive, RE-minerals monazite and xenotime contain major amounts of P2O5 (25-35%, Sarbajna et al., 2007), phosphate can be obtained as a co-product during their mineral processing for REEs.

### Recovery of CMs from RMs by Ore Dressing and Extractive Techniques

As majority of the CMs are in major contents in their respective, usually heavy and/or magnetic minerals in diverse radioactive materials (RMs), e.g., LREEs in monazite and HREEs in xenotime in mineral sands, recovery of CMs from RMs should start with the identification of specific minerals for their contained

CMs. This is to be followed by a general, sequential procedure involving (i) separation and concentration of such minerals from their RMs, using the mineral beneficiation techniques of gravity (e.g., hydro-cyclones and spirals), magnetic (e.g., low and high intensity magnetic separators) and electrostatic (e.g., high tension separator) separations, which are based on the physical parameters of minerals, such as density, magnetic susceptibility and conductivity; (ii) extracting next the CMs by techniques like hydro-/pyro-metallurgy and bio-/nano-technology and (iii) purification of different CMs in a smelter/refinery. In India, the Indian Rare Earths Ltd. (IREL) is processing placer monazite for value-added products like RE-chloride, RE-Th oxide cake and Th-concentrate, and the composite REchloride is further processed to obtain high purity oxides of La, Pr, Nd, Sm-Gd and Y, and Ce-carbonate. As the REs in monazite comprise basically LREEs, IREL is interested in sourcing HREEs and is willing to collaborate with foreign companies in this quest (Patra, 2014). In Australia, the mineral sand industry major, Iluka is planning to produce separated rare earth oxides at its rare earth refinery at the Eneabba mine site in Western Australia, with its Wimmera project serving as long-life rare earth concentrate feed source to the above refinery (Australian Mining, Jan. 12, 2022). Some important CMs like the battery metals, Co and Ni along with Cu and Mo are recoverable from sulphides in RMs. For example, the Rohil-Ghateswar multi-metal U-ore (~ 0.04% U3O8, 0.14% Cu and 0.024% Mo) in arid Rajasthan contains the above four CMs in the sulphides of pyrite, pyrrhotite, chalcopyrite, molybdenite and covellite. Sreenivas et al., (2018) have undertaken the process development studies to recover multi-metals from a composite feed of this sample, for which four options have been formulated, out of which the fourth 'Comminution - separation of option comprising ferromagnetic pyrrhotite by magnetic separation hydrometallurgical recovery of U – gravity separation of leach residue for non-magnetic heavy sulphide minerals recovery - tailings disposal' is found to be the best due to maximum recovery of both U (80-83%) and byproducts of Cu and Mo (~ 75% at the pre-concentration stage) with minimum water inventory due to easy recyclability of water, used in both the stages of magnetic and gravity separation. The sulphide mineral concentrate comprising Cu, Mo and Fe with traces of Ni and Co was subjected to 'roasting-leaching process' for quantitative separation of Cu, Mo, Ni and Co, while keeping the Fe-oxide phases as an insoluble fraction. Roasting converts the sulphides of Cu, Ni and Co to their respective sulphates and transforms the sulphides of Mo and Fe to their respective oxides by careful control of the roasting temperature. The sulphates of Cu, Ni and

Co are soluble in mild acidic aqueous medium and MoO in alkaline medium, whereas FeO is insoluble. A forward integration approach helped in treating lowgrade concentrates for maximum overall recovery of Cu and Mo Sreenivas et al., (2018). These examples illustrate the recoverability of CMs in RMs. It is desirable that more such studies are required for recovering CMs from RMs, followed by their purification by appropriate methods of chemical-/biochemical-/nano-technology, for which extensive R&D is to be carried out. By this swadeshi (indigenous) way under the Prime Minister's Make in India programme, the Country's requirement of critical/strategic minerals can be met to a notable extent, besides value-addition during the processing of RMs. Lastly, it is suggested that the AMD should actively work for the CMs recoverable from the RMs, as a part of their presently mandated exploration for radioactive (atomic) minerals.

### CONCLUSIONS

Critical Minerals (CMs) are (i) mostly metals and a few non-metals, (ii) ranked as most critical for the world's major industrial economies, (iii) vital for the economic well-being of the Society (iv) extensively used in (a) futuristic developments and (b) diverse conventional, high-tech and cutting-edge technologies; but their supply may be at risk due to several factors. The list of CMs is dynamic and depends upon their availability and perspective of each country.

From the point of availability in and perspective of India, 34 CMs were identified. Of these, > 60% is recoverable from diverse radioactive materials (RMs), and their important industrial uses are presented. Many Indian examples of the CMs recoverable from diverse RMs are given and discussed. A brief account on the methods of recovery, extraction and purification of these CMs with a few examples is presented. The need for further work, including R & D, in this field is highlighted, which may be taken up under the Prime Minister's Make in India programme for swadeshi supply of these CMs to meet much of their demand within the Country, besides value-addition during the processing of RMs.

#### ACKNOWLEDGEMENTS

A synthesis of this article, in the form of 3-slides under the poster session – 4, was presented in the National Seminar on '75 years of Mineral Exploration and Future Challenges in India' (MEFCI-2022, April 5-6, 2022), organized jointly by AMD, Hyderabad and Geological Society of India, Bengaluru. My sincere thanks are due to (i) the organizers of MEFCI-2022 and (ii) my former colleagues in AMD for their help and support during my service in AMD.

#### REFERENCES

- [1] Australian Government, Dept. of Industry, Science, Energy and Resources. Investing in critical minerals in Australia, https://www.industry.gov.au
   > invest
- [2] Australian Mining (2022) Iluka gets rare referral win. January 12, 2022, australianmining.com.au, https://www.australianmining.com.au/new/ilukagets-rare-referral-win/
- [3] Chandrasekaran, S. and Murugan, C. (2001) Heavy minerals in the beach and coastal red sands (Teris) of Tamil Nadu. AMD's EARFAM, v. 14, pp. 87-109.
- [4] Critical Minerals. Geoscience Australia; https://www.ga.gov.au resources;
- [5] Dept. of the Interior, US Geological Survey (2018) Critical minerals in the Federal Register. Interior releases 2018's final list of 35 minerals deemed critical to U.S., National Security and economy.
- [6] Dhana Raju, R. (2019) Sandstone type U-deposits at Domiasiat – Wahkyn and U-occurrences, Meghapaya, Book-Chapter 7, pp. 394-469, In: Indian Uranium Deposits, Cambridge Scholars Publishing, Newcastle upon Tyne, UK, ISBN (10): 1-5275-4046-4, ISBN (13): 978-1-5275-4046-0, xxvi + 535 p.
- [7] Dhana Raju, R. (2020) Critical minerals: their nature, occurrence, recovery and uses. Curr. Sci., v. 119 (6), 25 September 2020, pp. 919-925.
- [8] Dhana Raju, R. (2021) (Ed.) Indian Placer Deposits. Cambridge Scholars Publishing, Newcastle upon Tyne, UK, ISBN (10): 1-5275-6933-0, ISBN ((13): 978-1-5275-6933-1, xxxv + 621 p.
- [9] Dhana Raju, R., Ali, M.A. and Krishnan, S. (Eds.) (2001). Spl. Isssue on "Beach and Inland Heavy Mineral Sand Deposits of India'. AMD's EARFAM, v. 13, 159 p.
- [10] Dhana Raju, R., Kumar, M.K., Babu, E.V.S.S.K. and Pandit, S.A. (2002) Uranium mineralization in the Neoproterozoic Bhima Basin at Gogi and near Ukinal: An Ore Petrological Study. Jour. Geol. Soc. India, v. 59 (4), pp. 299-321.
- [11] Dhana Raju, R., Panneer Selvam, A. and Virnave, S.N. (1989) Characterization of the Upper Cretaceous Lower Mahadek Sandstone and Its Uranium Mineralization in the Domiasiat-Gomaghat-Pdengshakap area, Meghalaya, India. AMD's EARFAM, v. 2, pp.1-27.
- [12] Dhana Raju, R., Vijayalakshmi, S., Mahalingam, T.R., Rao, C.R.M. and Roy, Madhuparna (1998)
   Platinum Group Elements in REE-Th-Zr-Ti-rich

tremolite-actinolite rock from Vattalakki, Palghat district, Kerala. Jour. Geol. Soc. India, v. 51 (4), pp. 523-526.

- [13] Dwivedy, K.K. (1996) Characterisation and beneficiation of pegmatitic rare minerals in parts of the Bastar and Koraput districts, Madhya Pradesh-Orissa. Ph.D. thesis (Unpublished), Osmania University, Hyderabad, 188 p.
- [14] Federal Register (The Daily journal of the United States Govt.) 2021 Draft List of Critical Minerals [FR Doc. 2021-24488 Filed 11-8-21; 8:45 am].
- [15] IAEA (2001) Mining and milling of uranium ore: Indian scenario, Impact of New Environmental and Safety Regulations on Uranium Exploration, Mining, Milling and Management of its Waste. IAEA-TECDOC-1244, Vienna, pp 307-331.
- [16] Indian Minerals Year Book (2019) (Part III: Mineral Reviews, 58th ed., ilmenite and rutile) Indian Bureau of Mines, Ministry of Mines, Govt. of India, Sept. 2020.
- [17] Jagannadha Rao, M., Raju, U.P.N., Raja Rao, G. and Mounika, K.S.S. (2018) Beach placer deposits of India, their distribution, mineralogy and sustainable-mining with reference to placer ilmenite. Intl. Jour. Sci. and Res. (IJSR), v.7, pp. 1437-1443.
- [18] Krishna, K.V.G. and Thirupathi, P.V. (1999) Rare Metal and Rare Earth Pegmatites of Southern India.
  In: Mahadevan, T.M. and Dhana Raju, R. (Eds.), Spl. Issue on 'Rare Metal and Rare Earth Pegmatites of India', AMD's EARFAM, v. 12, 133-167.
- [19] Krishnan, S., Viswanathan, G. and Balachandran, K. (2001) Heavy Mineral Sand Deposits of Kerala. (Eds.) R. Dhana Raju, Mir Azam Ali and S. Krishnan, AMD's Spl. Issue on "Beach and Inland Heavy Mineral Sand Deposits of India", EARFAM, v. 13, pp. 111-146.
- [20] Madhuparna Sarkar, Ramprasad, V., Murthy, D.S.R. and Dhana Raju, R. (1990) Rare-Earths, Thorium, Zirconium, and Titanium-rich Tremolite-Actinolite Rock from Vattalakki, Palghat district, Kerala, India: A preliminary petrographic and geochemical study. AMD's EARFAM, v. 3, pp. 131-140.
- [21] Mahadevan, T.M. and Dhana Raju, R. (Eds.) (1999)
   Rare Metal and Rare Earth pegmatites of India.
   AMD's Spl. Issue, EARFAM, v. 12, 171 p.
- [22] Maithani, P.B. and Nagar, R.K. (1999) Rare Metal and Rare Earth Pegmatites of Western India. In: Mahadevan, T.M. and Dhana Raju, R. (Eds.) (1999) Rare Metal and Rare Earth pegmatites of India. Spl. Issue, AMD's EARFAM, v. 12, pp. 101-131.

- [23] Mohanty, A.K., Das, S.K., Vijayan, V., Sengupta, D. and Saha, S.K. (2003) Geochemical studies of monazite sands of Chhatrapur beach placer deposit of Orissa, India by PIXE and EDXRF methods. Nuclear Instruments and Methods in Physical Research, B 211, pp. 145-154.
- [24] Padhi, A.K., Aravind, S.L., Saxena, A., Kumar, K., Choudhury, D.K., Purohit, R.K., Nanda, L.K. and Rai, A.K. (2016). Uranium potential of Mesoproterozoic North Delhi Fold Belt, Rajasthan, India: An Overview. AMD's EARFAM, v. 26, pp. 53-70.
- [25] Panda, N.K., Rajagopalan, V. and Ravi, G.S. (2003) Rare Earth Element Geochemistry of placer monazites from Kalingapatnam coast, Srikakulam district, Andhra Pradesh. Jour. Geol. Soc. India, v. 62(4), pp. 429-438.
- [26] Panda, N.K., Sahoo, P., Rao, A.Y., Ramesh Kumar, K. and Rai, A.K. (2017) Concentration and distribution of rare earth elements in beach placer garnets of Kalingapatnam coast, and their potential for heavy rare earths, Andhra Pradesh, India. Jour. Geoscience. Res., v. 1, pp. 131-138.
- [27] Patra, R.N. (2014) Latest scenario in rare earth and atomic minerals in India. IREL PDAC\_2014.pdf.
- [28] President of the United States. Executive order 13817 pdf, February, 2018; https://www.whitehouse.gov.
- [29] Ramesh Babu, P.V. (1999) Rare Metal and Rare Earth Pegmatites of Central India. In: Mahadevan, T.M. and Dhana Raju, R. (Eds.), Spl. Issue on 'Rare Metal and Rare Earth Pegmatites of India', AMD's EARFAM, v. 12, pp. 7-52.
- [30] Ramesh Babu, P.V. (2021) Rare Earth Riverine Placer deposits in Eastern India, book chapter, pp. 200-251. In: (Ed.) R. Dhana Raju, Indian Placer Deposits, Cambridge Scholars Publishing, Newcastle upon Tyne, UK, ISBN (10): 1-5275-6933-0, ISBN ((13): 978-1-5275-6933-1, xxxv + 621 p.
- [31] Ranadive, K and Jawadand (2019) Strategic minerals in India: present status and future challenges. Miner. Econ., v. 32, pp. 337-352.
- [32] Rao, N.K. and Rao, G.V.U. (1980) Uraninite in the Uranium deposits of Singhbhum Shear Zone, Bihar. Jour. Geol. Soc. India, v. 21, pp. 387-397.
- [33] Rao, D.S. and Sengupta, D. (2014) Electron microscopic studies of ilmenite from Chhatrapur coast, Odisha, India their implications in processing. Jour. Geochem., Article ID 192639, http://dx.doi.org/10.1155/2014/192639, pp. 1-8.
- [34] Ravi, G.S. (2021) Shoreline and inland heavy mineral sand deposits of Odisha, eastern India. In:

(Ed.) R. Dhana Raju, Indian Placer Deposits, Cambridge Scholars Publishing, Newcastle upon Tyne, UK, Book Chapter 10, pp. 462-564.

- [35] Roy, Minati and Dhana Raju, R. (1999) Mineragraphy and EMP study of the U-phases and pyrite of the dolostone-hosted U-deposit at Tummalapalle, Cuddapah district, Andhra Pradesh and its implication on genesis and exploitation. Jour. Appl. Geochem., v. 1(2), pp. 53-75.
- [36] Sarbajna, C., Paul, A.K. and Rajagopalan, V. (2007) Mineral chemistry of columbite-tantalite, monazite and xenotime from Siri River placers, Jashpur district, Chhattisgarh. Gond. Geol. Mag. Spl. Vol. 10, pp.161-171.
- [37] Sarbajna, C., Sinha, R.P., Krishnamurthy, P., Krishna, K.V.G., Viswanathan, R. and Banerjee, D.C. (1999) Mineralogy and geochemistry of alkali beryl from rare metal bearing pegmatites of Marlagalla-Allapatna, Mandya district, Karnataka. Jour. Geol. Soc. India, v.54 (6), pp. 599-608.
- [38] Sen, D.B., Sachan, A.S., Padhi, A.K. and Mathur, S.K. (2002) Uranium exploration in the Cretaceous Mahadek sediments of the Meghalaya Plateau. EARFAM, v. 14, pp. 29-58.
- [39] Singh, Govind, Singh, Rajendra, Sharma, D.K., Yadav, O.P. and Jain, R.B. (1998) Uranium and REE potential of the Albitite-Pyroxenite-Microclinite belt of Rajasthan, India. AMD's EARFAM, v. 11, pp. 1-12.
- [40] Singh, H., Rao, M.S., Reddy, N.V.K., Padmanabhan, N.P.H., Iyer, N.V. and Rao, G.V.U. (1979) Oxidative roasting of molybdenite and volatilization of thorium in a multiple hearth furnace. BARC Project Report, pp. 1-566.
- [41] Singh, Rajendra, Dhana Raju, R. and Singh, A.K.
  (1994) Radioactive Fossil Placer (Cyrtolite-Hematite Ironstone) from the Boleng area, East Siang district, Arunachal Pradesh, India. Jour. Geol. Soc. India, v. 43 (4), pp. 407-413.
- [42] Singh, Yamuna (2020) Rare Earth Element Resources: Indian context, Springer, 410 p., https://books.google.com
- [43] Sreenivas, T., Patel, A.B., Ram Karan, Anandarao, K., Rajan, K.C. and Madangopal, K. (2018) Comprehensive extraction scheme for multi-metal recovery from metasomatite-albitite hosted low grade Indian uranium ore. In: Book of abstracts and extended abstracts of the Intl. Symp. on 'Uranium Raw Material for Nuclear Fuel Cycle: Exploration, Mining, Production, Supply, Demand, Economics and Environmental Issues. (URAM 2018), IAEA, June 25-29, Vienna, Austria. Extended Abst., pp. 399-401.